

# Glass Structure and First-Principles NMR

Thibault Charpentier

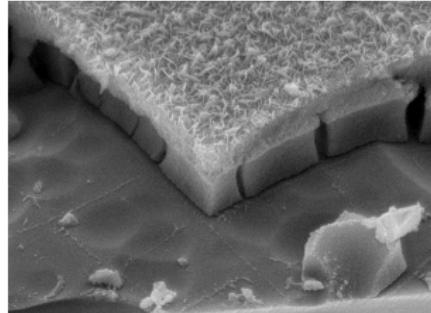
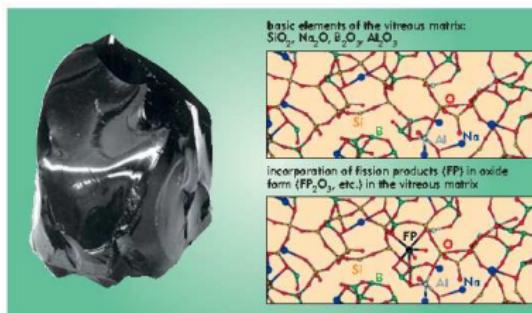
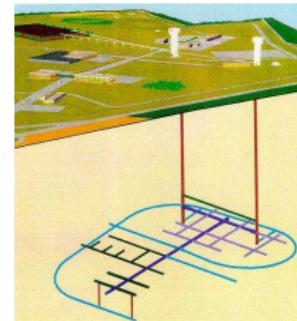
CEA / IRAMIS / SIS2M / LSDRM

CECAM Tutorial - September 2009



# Glass: a Nuclear Waste Storage Material

Long term behavior studies: structure and alteration



# R7T7 & SON68 Glasses

A complex borosilicate glass comprising more than 30 oxides

oxide	% (w)
$^{29}\text{SiO}_2$	45.12
$^{27}\text{Al}_2\text{O}_3$	4.87
$^{11}\text{B}_2\text{O}_3$	13.92
$^{6,7}\text{Li}_2\text{O}$	1.97
$^{23}\text{Na}_2\text{O}$	9.78
$^{43}\text{CaO}$	4.01
$\text{ZrO}_2$	0.99
$\text{ZnO}$	2.48
$\text{Fe}_2\text{O}_3$	2.89
$\text{P}_2\text{O}_5$	0.28
$\text{NiO}$	0.41
$\text{Cr}_2\text{O}_3$	0.50
Fission Products	10.35
Actinides	0.89
Platinoides	1.54

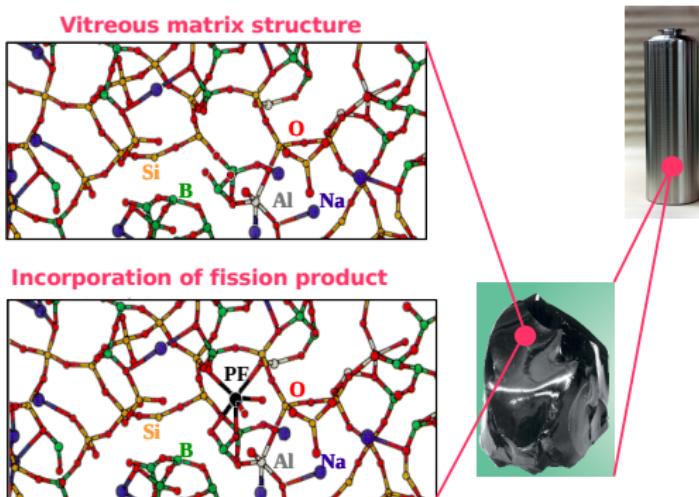


For Solid State NMR studies:  
Simplified Composition  
3-8 oxides  
NMR Probes

$^{11}\text{B}$   $^{27}\text{Al}$   $^{29}\text{Si}$   $^{23}\text{Na}$   $^{43}\text{Ca}$   $^{6,7}\text{Li}$   $^{17}\text{O}$

# NMR Structural studies

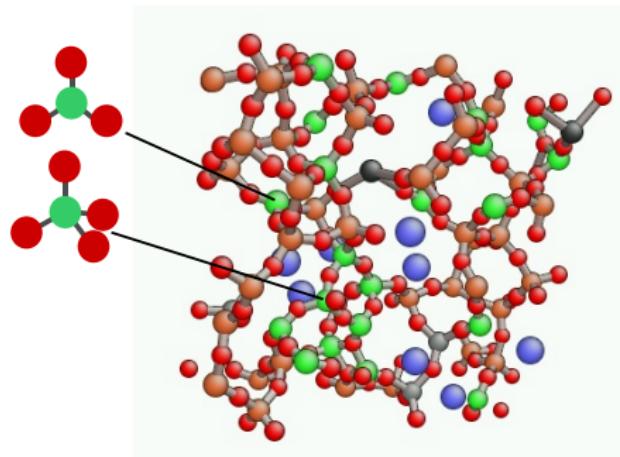
Probing the glass structure at the atomic scale



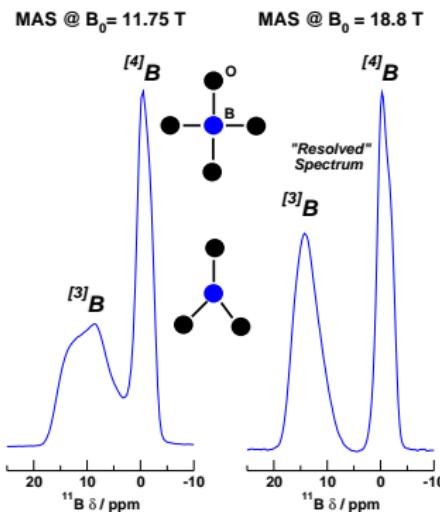
- Glass accommodates the elements present in the spent fuel
- FP atoms are an integral part of the glass

# Solid State Nuclear Magnetic Resonance Spectroscopy

## Boron speciation in borosilicate glass



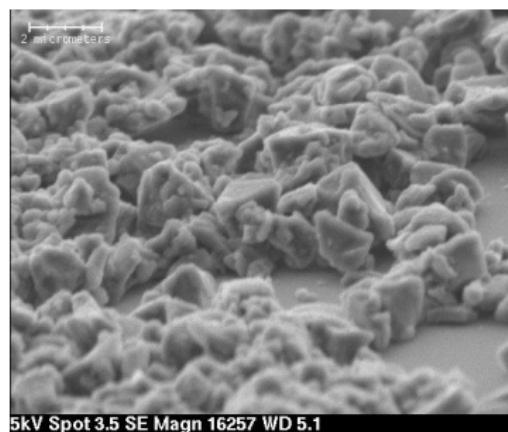
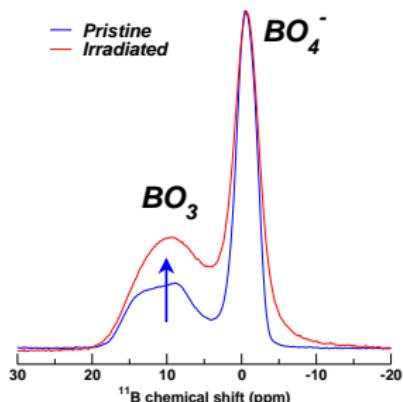
Identifying the structural units of the glass network



High Field MAS NMR:  
Boron speciation resolved

# Solid State Nuclear Magnetic Resonance Spectroscopy

Heavy Ion Damage: amorphisation of amorphous material ?

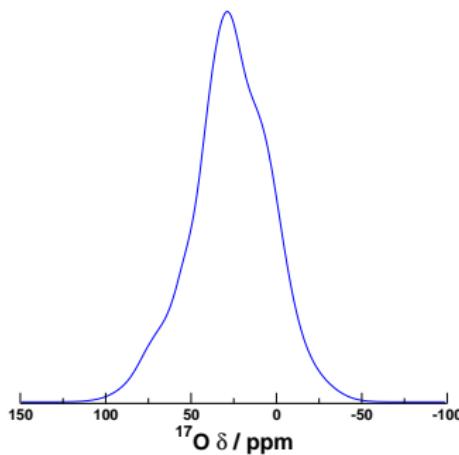


Coll. CEA/DEN/DTCD S. Peugeot, J.M. Delaye

# Oxygen Speciation in borosilicate glass

Reading the silicate network structure

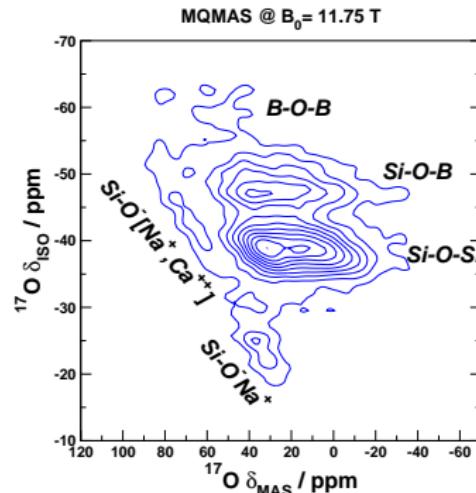
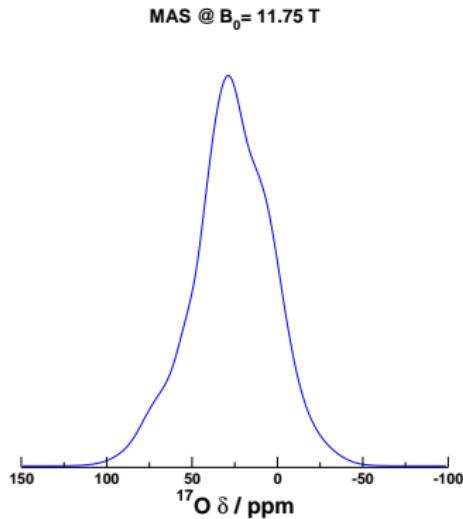
MAS @  $B_0 = 11.75$  T



MAS Unresolved

# Oxygen Speciation in borosilicate glass

Reading the silicate network structure

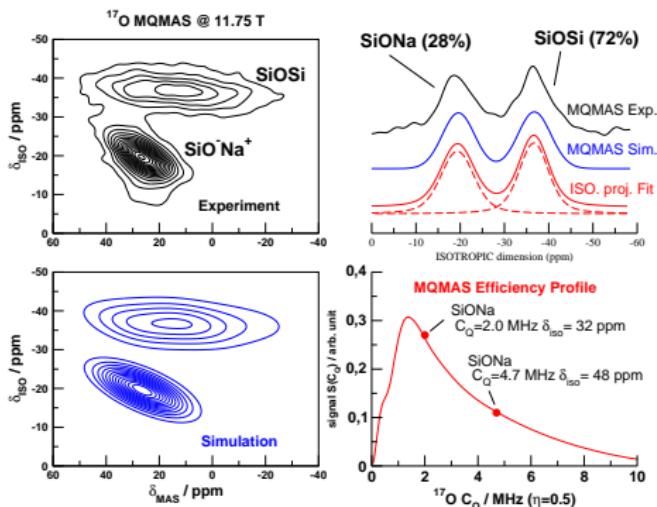


2D technique required: MQMAS

# Modeling and Quantifying $^{17}\text{O}$ MQMAS Spectroscopy

Quantification & NMR Distribution

F. Angeli, T. Charpentier et al. JNCS 2008



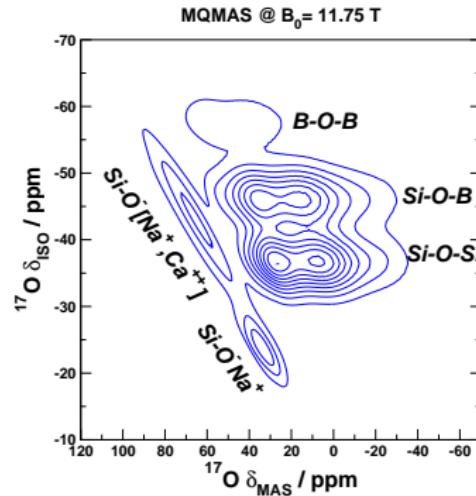
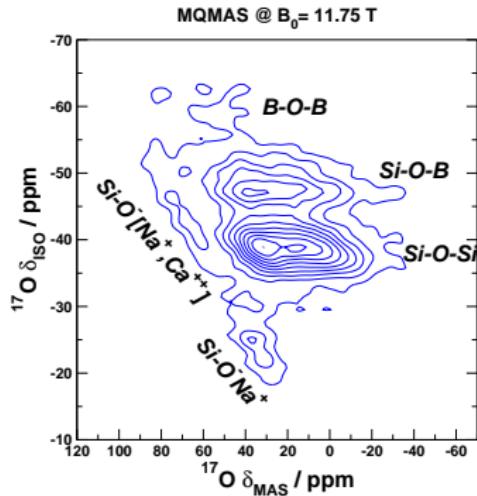
Uncorrelated model of NMR parameter distribution

$$p(C_Q, \eta, \delta_{\text{iso}}) = G(C_Q) \times G(\eta) \times G(\delta_{\text{iso}})$$

# Oxygen Speciation in borosilicate glass

Reading the silicate network structure

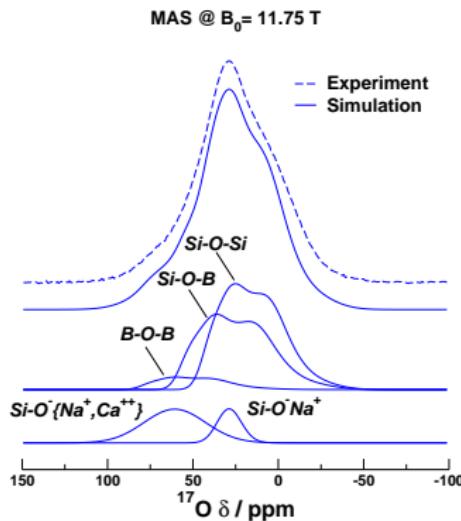
F. Angeli, T. Charpentier et al., JNCS 2008



*Total Simulation of MQMAS spectroscopy*

# Oxygen Speciation in borosilicate glass

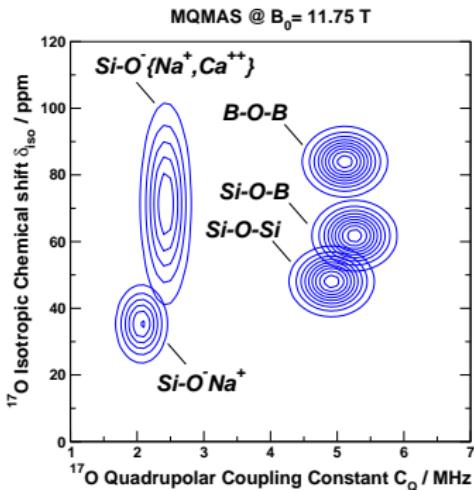
Quantifying the glass network structure



## Glass Topology

- Bridging Oxygen Atoms:  
Si-O-Si, Si-O-B, B-O-B  
Glass homogeneity at nanoscale
- Non Bridging Oxygen: Si-O<sup>-</sup>Na<sup>+</sup>  
Glass Polymerization degree
- Non Bridging Oxygen:  
Si-O<sup>-</sup>(Na<sup>+</sup>,Ca<sup>++</sup>)  
Evidence of a Mixed Alkali -  
Alkaline Earth Effect

# Quantifying glass topological disorder



## Glass Topology

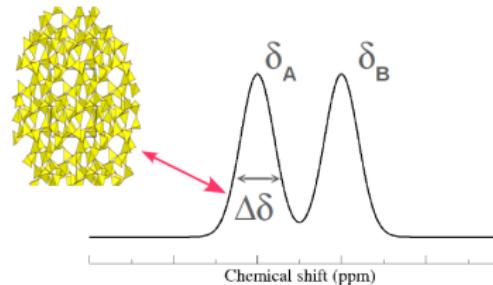
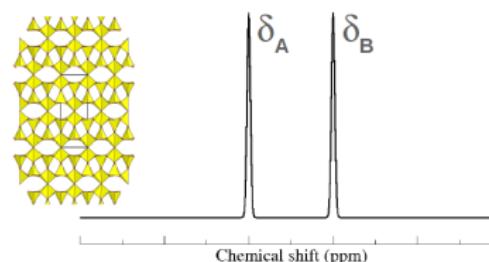
- Reconstruction of the NMR parameter distribution  $\Pi$
- Correlating the *local disorder* to the NMR spectrum line shape ?
- $\Pi(\text{NMR}) \Rightarrow \Pi(\text{Structure})$  ?

# NMR of disordered Materials

## Modeling and Interpreting

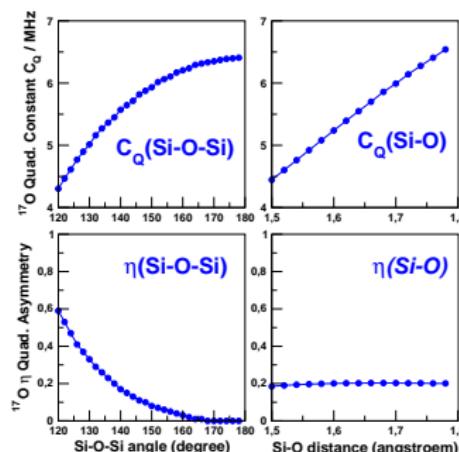
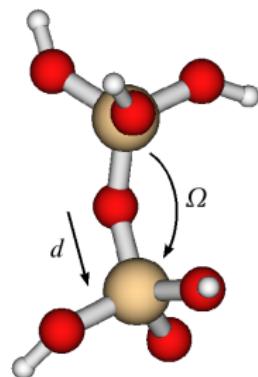
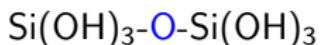
A fundamental question:

Correlating the local disorder to the NMR spectrum line shape



# Understanding NMR / Structure relationships

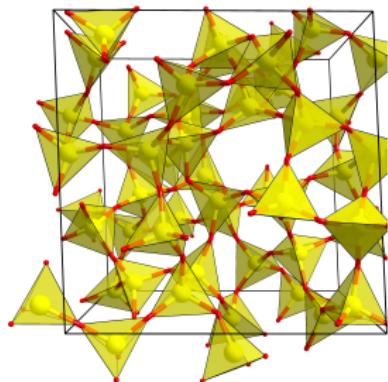
Quantum Mechanical calculations: the Cluster Approach



$^{17}\text{O}$   $C_Q$  and  $\eta_Q$  NMR parameters are almost exclusively controlled by local properties: (Si-O-Si bond angle and Si-O bond length)

# Combining MD simulations with fp NMR calculations

GIPAW: C.J. Pickard & F. Mauri, PRB 2001

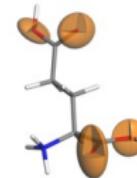


## GIPAW

### A solid state theory of Magnetic Resonance

- Home
- Theory
- Publications
- Codes:
  - PWSCF
  - CASTEP
- Links

GIPAW (Gauge Including Projector Augmented Waves) is a DFT based method to calculate magnetic resonance properties, exploiting the full translational symmetry of crystals. The use of pseudopotentials and plane waves provides an excellent balance of speed and accuracy.

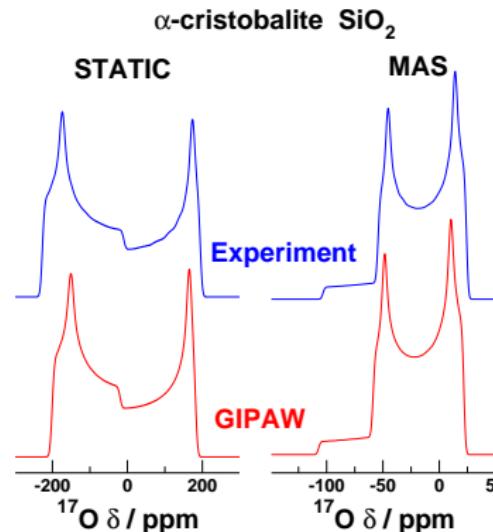


# The GIPAW method

Gauge Including Projector Augmented Wave

C.J. Pickard & F. Mauri, PRB 2001

- DFT using GGA (PBE) or LDA functionals.
- Plane Waves Expansion ( $e^{-ik \cdot r}$ )
  - 3D FFT, Parallel Code
  - Periodic Boundary Conditions
- Pseudopotential approximation of core electrons
- GI-PAW
  - PAW: Reconstruction of the wave function at the nucleus
  - GI: Gauge Invariance

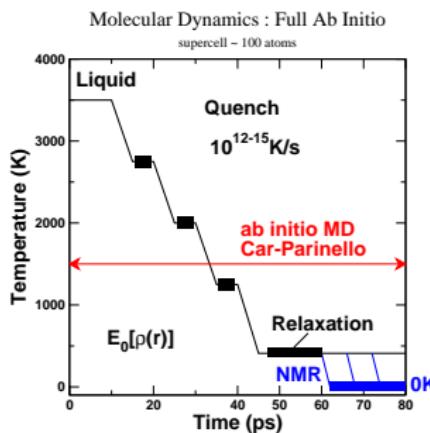
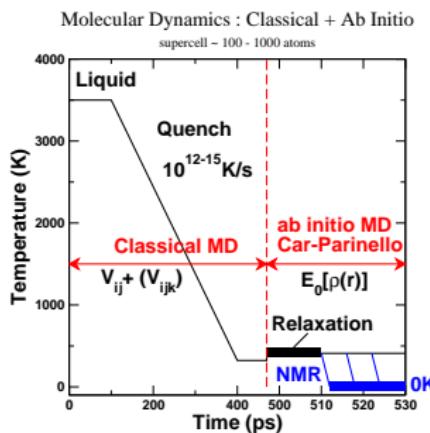


Exp.: Spearing et al. Phys. Chem. Minerals 1992.

Accuracy: GIPAW outperforms all previous approaches

# Molecular Dynamics

## MD-GIPAW Methodology



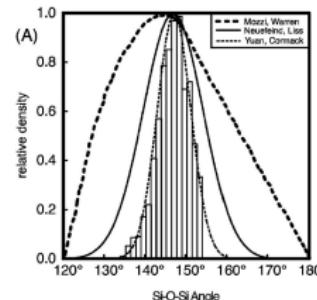
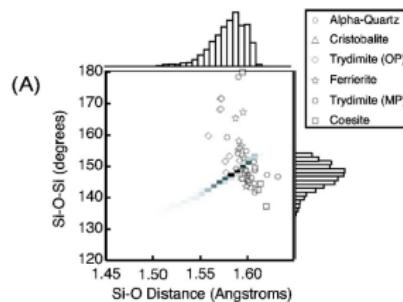
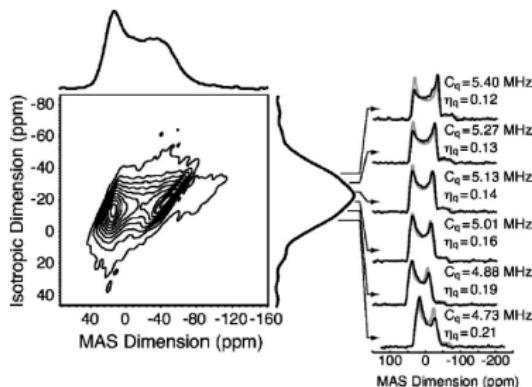
MD: Collaboration S. Ispas, P. Kroll, G. Ferlat, F. Mauri, J.M. Delaye

# The controversy: Bond Angle Distribution in vitreous silica

## Revisiting the interpretation of $^{17}\text{O}$ NMR data

PHYSICAL REVIEW B **70**, 064202 (2004)

## Correlated structural distributions in silica glass

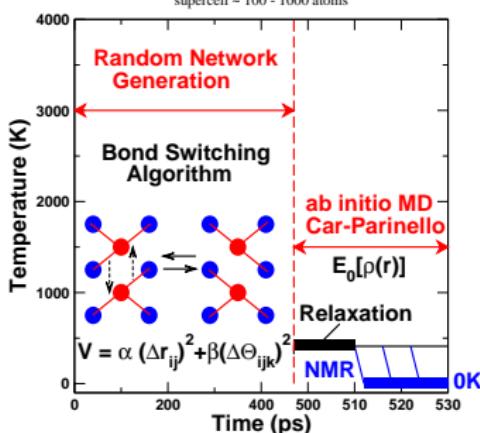
Ted M. Clark and Philip J. Grandinetti\*  
Department of Chemistry, The Ohio State University, 120 West 18th Avenue, Columbus, Ohio 43210-1173, USAPierre Florian  
CNRS-CRMRT, 1D Av. de la Recherche Scientifique, 45071 Orléans Cedex 2, FranceJonathan F. Stebbins  
Department of Geological and Environmental Sciences, Stanford University, Stanford, California 94305-2115, USA  
(Received 27 February 2004; revised manuscript received 4 June 2004; published 18 August 2004)

A Monte Carlo / MD Hybrid Approach

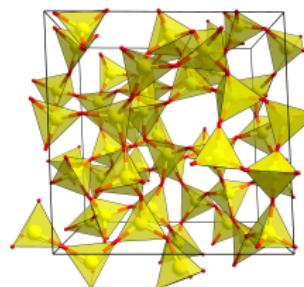
## Modeling Glass Structure: Bond Switching Algorithm

Molecular Dynamics : Monte Carlo + Ab Initio

supercell = 100–1000 atoms

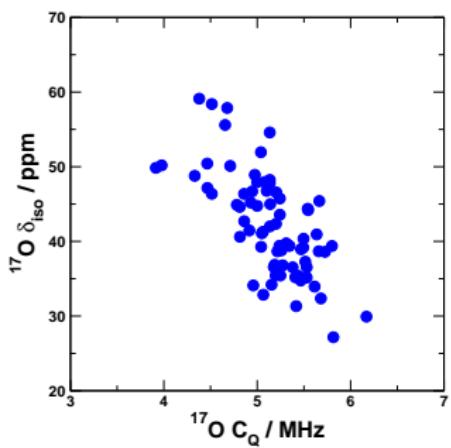


# Defect Free Vitreous Silica Continuous Random Network



# Simulation of NMR Spectra

The simple approach

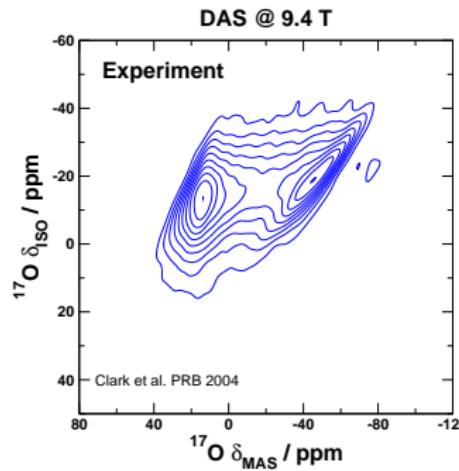
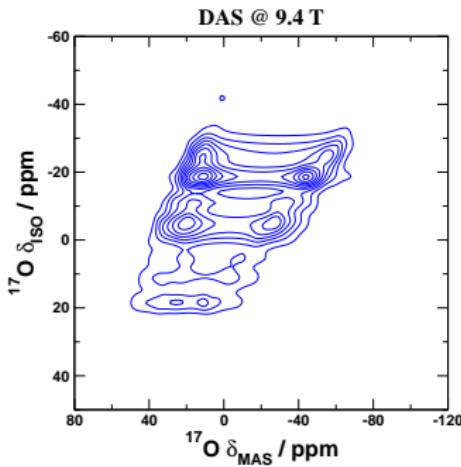


$$I(\vec{\nu}) = \sum_i I_{th}^i(\vec{\nu})$$



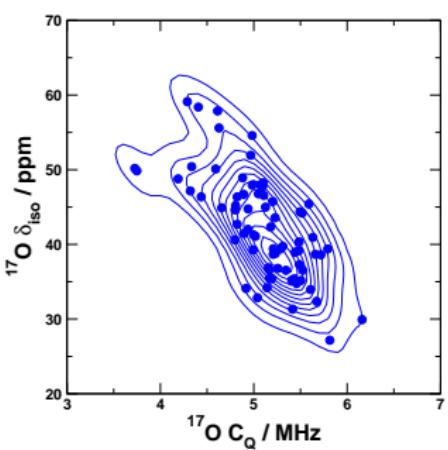
# Simulation of NMR Spectra

The simple approach



# Simulation of NMR Spectra

## Kernel Density Estimate Approach



$$\{C_Q, \eta, \delta_{iso}\}_i \Rightarrow p(C_Q, \eta, \delta_{iso})$$

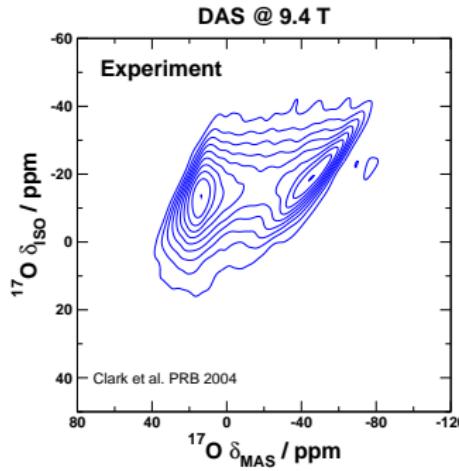
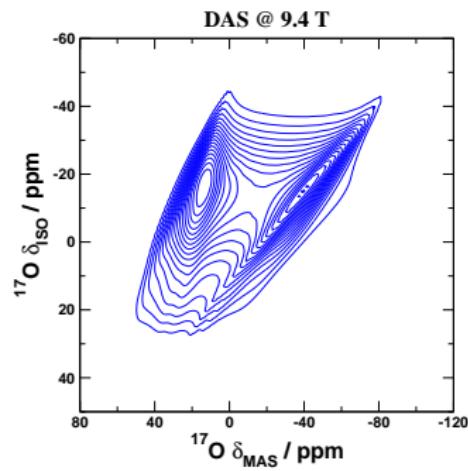
$$\text{Kernel: } p(\mathbf{x}) = \sum_i \mathcal{K}_{H_i} (\mathbf{x} - \mathbf{x}_i)$$

$$I(\vec{\nu}) = \int p(C_Q, \eta, \delta_{iso}) \\ \times I_{th}(\vec{\nu}; C_Q, \eta, \delta_{iso})$$



# Simulation of NMR Spectra

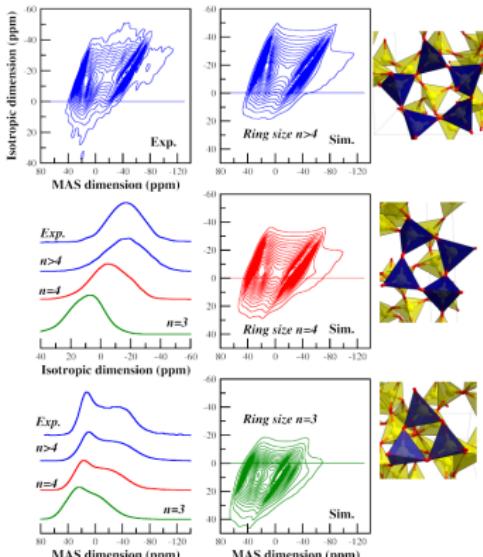
Kernel Density Estimate Approach



# Glass Topology

## The (small)-rings statistics

### $^{17}\text{O}$ NMR response of small rings



Ring Size	Local structure SiOSi degrees	$^{17}\text{O}$ NMR parameters		
		$C_Q$ MHz	$\eta$	$\delta_{iso}$ ppm
3	126.7 (4.1)	4.61 (0.35)	0.38 (0.15)	55.92 (7.85)
4	142.3 (10.9)	5.03 (0.39)	0.24 (0.13)	46.54 (6.42)
$> 4$	147.0 (12.2)	5.21 (0.41)	0.20 (0.12)	39.59 (6.55)
Experiment <sup>(a)</sup>		5.08 (0.37)	0.15 (0.04)	36.7 (4.3)

(a) Clark et al. PRB 2004

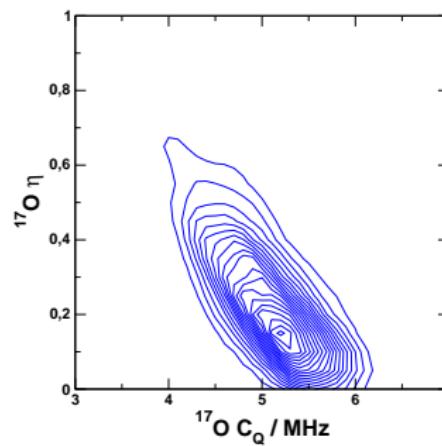
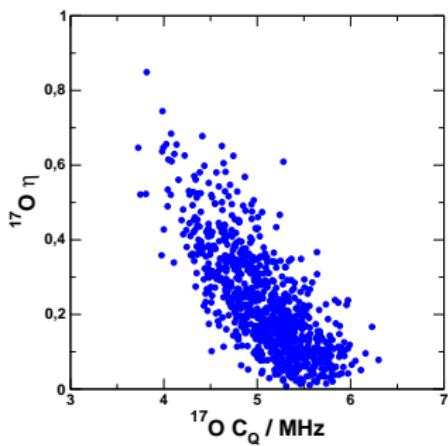
Ring Size	Local structure SiOSi degrees	$^{29}\text{Si}$ NMR parameters		
		$\delta_{iso}$ ppm	$\Omega_\sigma$ ppm	$\kappa_\sigma$
3	135.4 (12.4)	-97.97 (4.74)	31.27 (8.40)	0.10 (0.41)
4	144.0 (11.5)	-106.09 (5.52)	23.66 (8.25)	0.02 (0.40)
$> 4$	148.3 (12.3)	-110.31 (5.53)	17.50 (6.83)	-0.04 (0.41)
Experiment <sup>(a)</sup>		-111.2 (5.1)	n.a.	n.a.

(a) Clark et al. PRB 2004

T. Charpentier, P. Kroll, F. Mauri, J. Phys. Chem C, in press.

# Analysis of $^{17}\text{O}$ NMR parameter distribution

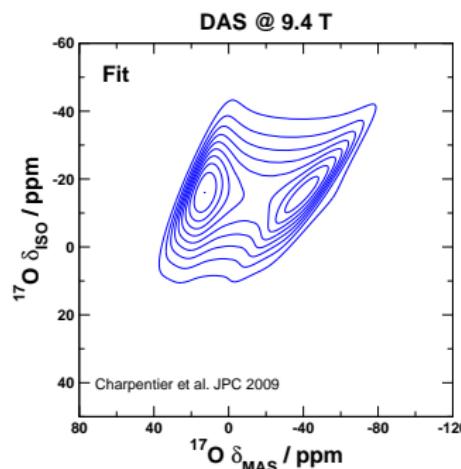
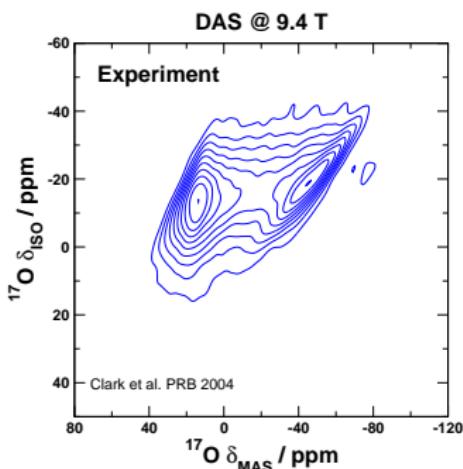
## Kernel Density Estimate Approach



Introduction of a correlated 3D NMR parameter distribution:

$$p(C_Q, \eta, \delta_{iso}) = G(C_Q - \bar{C}_Q) \times G(\eta - f_\eta(C_Q)) \times G(\delta_{iso} - f_\delta(C_Q))$$

# Quantitative Analysis of $^{17}\text{O}$ Experimental data



Experimental data can be well described in term of *disorder only*.

# Interpretation of $^{17}\text{O}$ Experimental data

## $^{17}\text{O}$ NMR parameter distribution

	Charpentier et al. JPC 2009		Clark et al. PRB 2004	
	Mean	Std. dev.	Mean	Std. dev.
$C_q$	5.07 MHz	0.453 MHz	5.08 MHz	0.372 MHz
$\eta_q$	0.157	0.095	0.150	0.0414
$\delta_{iso}$	36.5 ppm	7.55 ppm	36.7 ppm	4.30 ppm

# Interpretation of $^{17}\text{O}$ Experimental data

## $^{17}\text{O}$ NMR parameter distribution

	Charpentier et al. JPC 2009		Clark et al. PRB 2004	
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$C_q$	5.07 MHz	0.453 MHz	5.08 MHz	0.372 MHz
$\eta_q$	0.157	0.095	0.150	0.0414
$\delta_{iso}$	36.5 ppm	7.55 ppm	36.7 ppm	4.30 ppm

## Local geometry

	JPC 2009		PRB 2004		Diffraction.	
	Mean	Std. dev.	Mean	Std. dev.	Mean	Std. dev.
Si-O-Si	147.1°	11.17°	146.6°	3.78°	148.3°	7.5°
Si-O	1.60 Å	0.011 Å	1.58 Å	0.019 Å	1.61 Å	0.049 Å



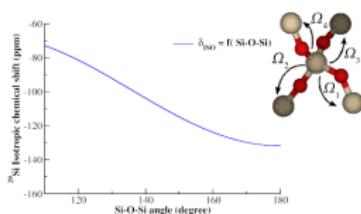
IRAMIS

# Bond Angle Distribution (BAD)

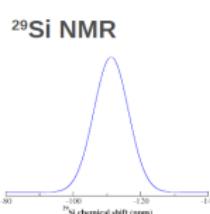
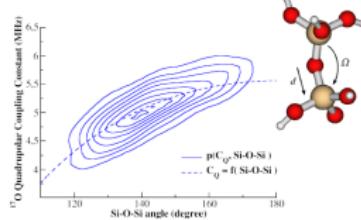
NMR / Structure relationships

T. Charpentier et al., J. Phys. Chem. C 2009

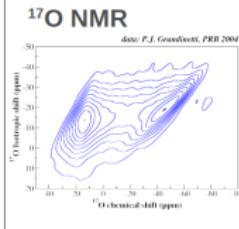
## Reconstruction of the Bond Angle Distribution (SiOSi) from NMR data



GIPAW + MD:  
NMR-Structure relationships



Structural Inversion

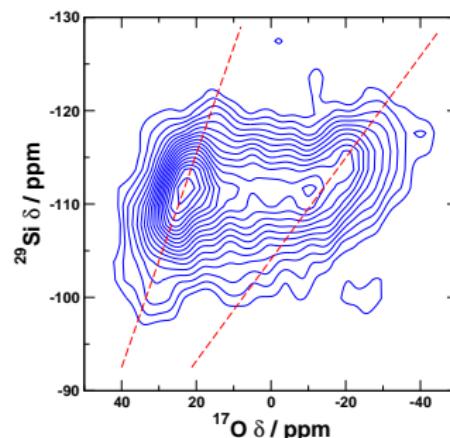
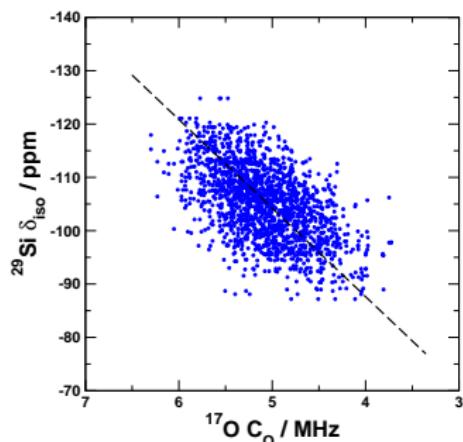


T. Charpentier, P. Kroll, F. Mauri, J. Phys. Chem. C, in press.

# Advanced NMR: Through Bond Correlation Spectroscopy

$^{29}\text{Si}$ - $^{17}\text{O}$  J-HMQC experiments.

GIPAW J Calc. coll. J. Yates Oxford

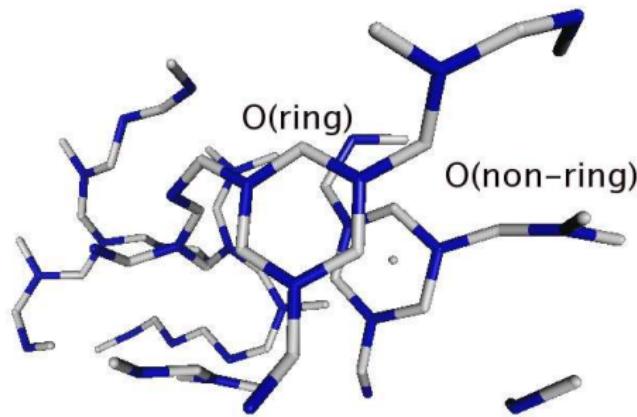


2D NMR enables the observation of spin pair  $^{29}\text{Si}$ - $^{17}\text{O}$  through the chemical bond (or through the space)

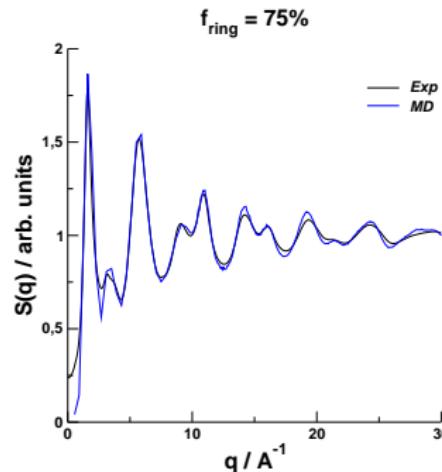
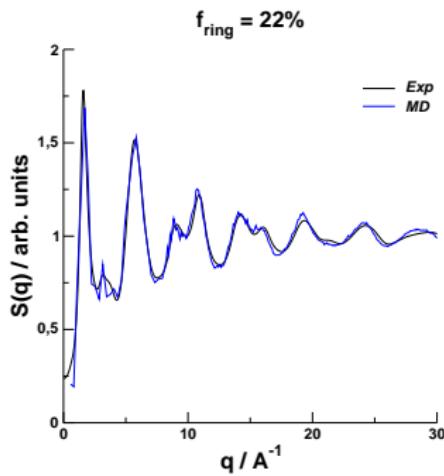
⇒ Observation of Bond Angle  $\theta_i - \theta_{i+1}$  correlation ...

# Topology of vitreous $B_2O_3$

G. Ferlat, T. Charpentier et al., PRL 2008

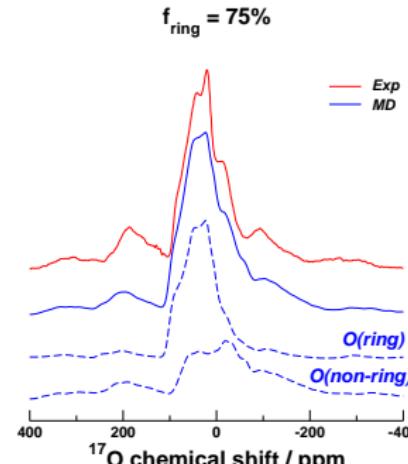
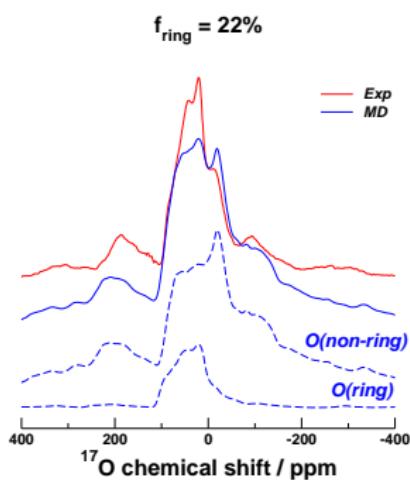


# Topology of vitreous $B_2O_3$



$S(q)$ : poor sensitivity to Bond Angle ?

# Topology of vitreous $B_2O_3$



NMR: high sensitivity to Bond Angle

# aiNMR: Perspectives

A new powerful tools to get more insight into NMR data

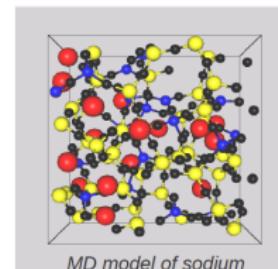
**Structural studies of complex glasses** (borosilicates, mixed alkali effect, Bioglass, ...)

**Dynamical effects**  
vibration, diffusion  $\Leftrightarrow$  relaxation

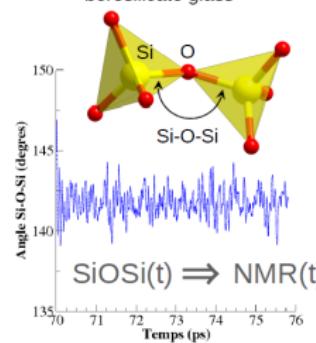
**Data modelling of Glass**  
(imitating bioNMR !)

Combining NMR constraints (from 1D, 2D, 3D) with diffraction and other spectroscopy (RAMAN, ...) to refine Glass structure models

*NMR Driven Reverse Monte Carlo*



MD model of sodium borosilicate glass



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