

# Density Functional Theory and the Local Density Approximation

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#### Outline

- Born-Oppenheimer approximation
- Hohenberg-Kohn theorems
- Kohn-Sham implementation
- The  $E_{xc}$  functional
- The local density approximation (LDA)
- Limits of current implementations of DFT



#### **Born-Oppenheimer Approximation**

 $\hat{H}_{N,e}\Psi_{N,e}(\{\mathbf{R}_{J}\},\{\mathbf{r}_{i}\}) = E_{N,e}\Psi_{N,e}(\{\mathbf{R}_{J}\},\{\mathbf{r}_{i}\})$ 

Electron mass much smaller than nuclear mass:

Timescale associated with nuclear motion much slower than that associated with electronic motion

Electrons follow instantaneously the motion of the nuclei, remaining always in the same stationary state of the Hamiltonian

$$\Psi_{N,e}(\{\mathbf{R}_J\}, \{\mathbf{r}_i\}) = \Theta_N(\{\mathbf{R}_J\}) \bigoplus_{\substack{\downarrow \\ \text{depends only parametrically on } \{\mathbf{R}_I\}}$$

Born-Oppenheimer approximation decouples the electronic problem from the ionic problem. The electronic problem is:

$$\hat{H} \Phi(\{\mathbf{r}_i\}) = E \Phi(\{\mathbf{r}_i\})$$

#### **The Electronic Problem**

Time-independent, non-relativistic Schroedinger equation in the Born-Oppenheimer approximation [in atomic units:  $e = m_e = \hbar = 1$ ]:

 $\hat{H} \Phi(\{\mathbf{r}_i\}) = E \Phi(\{\mathbf{r}_i\})$   $\hat{H} = \hat{T}_e + \hat{V}_{e-e} + \hat{V}_{N-e}$ where  $\hat{T}_e = -\frac{1}{2} \sum_{i}^{N} \nabla_i^2$   $\hat{V}_{e-e} = \sum_{i < j}^{N} \frac{1}{|\mathbf{r}_i - \mathbf{r}_j|}$   $\hat{V}_{N-e} = -\sum_{\alpha}^{N_{al}} \frac{Z_{\alpha}}{|\mathbf{r} - \mathbf{R}_{\alpha}|} \quad \text{External potential}$ The external potential and the number of electrons, N,

completely determine the Hamiltonian

Density functional theory (DFT) offers a solution ...

#### Density Functional Theory: the Theorems

Theorem 1.

The external potential is uniquely determined by the electronic charge density -  $n(\mathbf{r})$  - so the total energy is a unique functional of the density - E[n] !

Theorem 2.

The density which minimises the energy is the ground state density and the minimum energy is the ground state energy:

$$\operatorname{Min}_{n} E[n] = E_0$$

there is a *universal* functional *E*[n] which could be minimised to obtain the *exact* ground state density and energy.

Hohenberg & Kohn, PRB 136, 864 (1964)

#### What is the Functional?

$$E[n] = T_e[n] + V_{e-e}[n] + V_{ext}[n].$$

where  $T_e[n]$  is the kinetic energy and  $V_{e-e}$  is the electron-electron interaction.

 $V_{ext}[n]$  is trivial:

$$V_{ext}[n] = \int \hat{V}_{ext} n(\mathbf{r}) d\mathbf{r}$$
  
where  $\hat{V}_{ext} = -\sum_{\alpha}^{N_{at}} \frac{Z_{\alpha}}{|\mathbf{r} - \mathbf{R}_{\alpha}|}$ 

T and  $V_{e-e}$  need to be approximated !

### **E[n] – The Kohn Sham Approach**

Write the density in terms of a set of *N* non-interacting orbitals:

$$n(\mathbf{r}) = \sum_{i}^{N} \left| \phi_{i}(\mathbf{r}) \right|^{2}$$

The non interacting kinetic energy and the classical Coulomb interaction

$$T_{s}[n] = -\frac{1}{2} \sum_{i}^{N} \left\langle \phi_{i} \middle| \nabla^{2} \middle| \phi_{i} \right\rangle \qquad V_{H}[n] = \frac{1}{2} \int \frac{n(\mathbf{r}_{1})n(\mathbf{r}_{2})}{\left| \mathbf{r}_{1} - \mathbf{r}_{2} \right|} d\mathbf{r}_{1} d\mathbf{r}_{2}$$

This allows us to recast the energy functional as:

$$E[n] = T_{s}[n] + V_{ext}[n] + V_{H}[n] + E_{xc}[n]$$

where we have introduced the exchange-correlation functional:

$$E_{xc}[n] = (T[n] - T_s[n]) + (V_{e-e}[n] - V_H[n])$$

#### **The Kohn-Sham Equations**

Vary the energy with respect to the orbitals and ....

$$\left[-\frac{1}{2}\nabla^2 + V_{ext}(\mathbf{r}) + \int \frac{n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' + V_{xc}(\mathbf{r})\right] \phi_i(\mathbf{r}) = \varepsilon_i \phi_i(\mathbf{r})$$

where:

$$V_{xc}(\mathbf{r}) = \frac{\partial E_{xc}[n]}{\partial n(\mathbf{r})}$$

KS equations are solved via an iterative procedure until self-consistency is reached

No approximations, So...

If we knew  $E_{xc}[n]$  we could solve for the **exact** ground state energy and density !

Kohn & Sham, PRA 140, 1133 (1965)

#### **The Non-interacting System**

There exists an effective mean field potential which, when applied to a system of non-interacting fermions, will generate the exact ground state energy and charge density !!!



#### Density Functional Theory: The History



Thomas-Fermi-Dirac Model (1929): First model where the energy is expressed in terms of the electron density

E. Fermi, Nobel laureate 1938



Kohn-Hohenberg-Sham (1964): Exact univoque relationship between ground state energy and electron density

W. Kohn, Nobel laureate 1998

# **E**<sub>xc</sub>[n] - Properties

 $E_{xc}[n] = (T[n] - T_s[n]) + (V_{e-e}[n] - V_H[n])$ 

– Does not depend on  $V_{ext}$  (the specific system): it is a 'universal' functional.

– The exact dependence on  $n(\mathbf{r})$  is unknown

 $E_{xc}$  must be approximated in applications

#### Local Density Approximation (LDA)

The idea: build  $E_{xc}$  from the knowledge of the exchange-correlation energy per particle,  $\epsilon_{xc}$ , of the homogeneous electron gas

Kohn & Sham, PRA 140, 1133 (1965)

For the homogeneous electron gas the exact dependence of  $\epsilon_{xc}(n)$  can be computed by Quantum Monte Carlo.



# LDA to E<sub>xc</sub>: How it is built

$$E_{xc}^{LDA} = \int \varepsilon_{xc} (n(\mathbf{r})) n(\mathbf{r}) d\mathbf{r}$$



Picture courtesy of Andreas Savin

Kohn & Sham, PRA 140, 1133 (1965)

#### **LDA: Performance**

Kohn & Sham, PRA **140**, 1133 (1965): "We do not expect an accurate description of chemical bonding"

In reality:

- Covalent bonds
  Metallic bonds
  generally well described
- Ionic bonds

(tends to overbind slightly)

- Hydrogen bonds: LDA not adequate
- Van der Waals bonds: DFT not adequate but this is not just a LDA problem...

Why does LDA work in most cases ?

# **The Exchange-Correlation Hole**

The exchange correlation hole,  $P_{xc}(\mathbf{r}_1, \mathbf{r}_2)$ , is the probability of finding an electron at  $\mathbf{r}_2$  given that there is an electron at  $\mathbf{r}_1$ .

It is the hole the electron at  $\mathbf{r}_1$  digs for itself in the surrounding electronic density.

There are a number of properties which will be satisfied by the exact exchange correlation hole. For instance it should normalise to exactly one electron:

$$\int P_{xc}(\mathbf{r}_1,\mathbf{r}_2)d\mathbf{r}_2 = -1$$

LDA satisfies this rule.





# **DECISION OF COLSPANSION OF COLSPANDING OF COLSPANDIN**

0.000305

-0.00047

-0.00125

Hood et al PRB **57** 8972 (1998)

#### So Why does the LDA work ?

- Exact properties of the xc-hole maintained
- The electron-electron interaction depends only on the spherical average of the xc-hole – this is reasonably well reproduced
- The errors in the exchange and correlation energy densities tend to cancel

#### Some things that do not work in this approach

- Van der Waals interactions: due to mutual dynamical charge polarisation of the atoms → not properly included in any existing approximations to E<sub>xc</sub>
- Excited states: DFT is a ground state theory (ways forward: time-dependent DFT, GW, ...)
- Non Born-Oppenheimer processes (i.e, non-radiative transitions between electronic states)
- Self-interaction problem: each electron lives in the field created by all electrons including itself ( ways forward: SIC, hybrid DFT)





#### **Summary**

- Born-Oppenheimer approximation
- Hohenberg-Kohn theorems
- Kohn-Sham implementation
- The  $E_{xc}$  functional
- The local density approximation (LDA): how it is built, why it works in many cases
- Limits of current implementations of DFT (self interaction, van der Waals forces, ...)

#### <u>Next</u>

The magical world of electrons in crystals



#### **Useful References**

- Hohenberg & Kohn, PRB **136**, 864 (1964)
- Kohn & Sham, PRA 140, 1133 (1965)
- R. O. Jones and O. Gunnarsson, Rev. Mod. Phys. **61**, 689 (1989)
- M. C. Payne et al., Rev. Mod. Phys **64**, 1045 (1992)

• R. M. Martin, "*Electronic Structure: basic theory and practical methods"*, Cambridge University Press (2004)

• J. Kohanoff, "*Electronic Structure Calculations for Solids and Molecules: Theory and Computational Methods"*, Cambridge University Press (2006)